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Farzana Shaheen,^a Manzoor Ahmad, a Shazia Anjum, a* Azhar Abdul Rahman,^b Hoong-Kun Fun,^b Habib Ahmad, M. Igbal Choudhary a and Atta-ur-Rahmana

^aH. E. J. Research Institute of Chemistry, International Centre for Chemical Sciences, University of Karachi, Karachi 75270, Pakistan, ^bX-ray Crystallography Unit, School of Physics, Universiti Sains Malaysia, 11800 Penang Malaysia, and ^cJahanzeb Postgraduate College, Saidu Sharif, Swat, Pakistan

Correspondence e-mail: shazia.anjum@hej.edu

Key indicators

Single-crystal X-ray study T = 293 KMean $\sigma(C-C) = 0.007 \text{ Å}$ R factor = 0.063wR factor = 0.131 Data-to-parameter ratio = 7.8

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

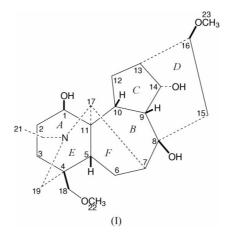
20-N-Ethyl-1,8,14-trihydroxy-16β,18dimethoxylycoctonine

The title compound, C₂₃H₄₇NO₅, has been isolated from Delphinium roylei for the first time. The molecular architechture comprises six fused rings, A-F. The six-membered rings, A and E, are in chair conformations, while ring D is in a distorted boat form. The seven-membered ring B is in a boat form and the remaining two five-membered rings, C and F, are in envelope conformations. The crystal structure is stabilized by $O-H\cdots O$, $C-H\cdots O$ and $C-H\cdots N$ interactions.

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Comment

Delphinium roylei Munz. is a 50-100 cm long perennial herb found at an elevation of 2300-2800 m (Ali & Nasir, 1991). Delphinium (Larkspur), an important genus of the family Ranunculaceae, is well known for its potential uses in medicine. The genus is recognized for its poisonous constitutents, which consist of biologically active and structurally complex diterpenoid and norditerpenoid alkaloids with febrifuge, sedative, cardiotonic and analgesic activities (Benn & Jacyno, 1983, 1984). Some of the species of Delphinium are reported to be used as insecticides and anti-rheumatics and for the treatment of sciatica (Baytop, 1984). Recently, our research group has reported anti-epileptic activity in aqueous and acetone fractions of *Delphinium denudatum* (Raza, Shaheen, Choudhary, Rafiq et al., 2001; Raza, Shaheen, Choudhary, Suria et al., 2001; Raza et al., 2003). During our ongoing phytochemical investigations on this plant, we have isolated and performed single-crystal X-ray diffraction studies on the title compound, also known as talatisidine, (I), which has previously been reported as being extracted from other species of the genus *Delphinium* (Pelletier et al., 1967).



Compound (I) crystallizes in the space group C2. The bond lengths and angles in (I) show normal values (Allen et al., 1987). The absolute stereochemistry could not be determined,

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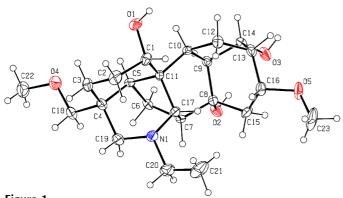


Figure 1
The structure of (I), showing 30% probability displacement ellipsoids and the atom-numbering scheme.

as the anomalous dispersion effects are too weak. However, the absolute configuration of (I) is assumed to be the same as that reported for talatisidine by Pelletier *et al.* (1967).

The molecule of (I) has a rigid structure consisting of six main rings, A–F. In the lycoctonine nucleus of (I), the junctions of the main rings A/E and B/C are cis [C17—C11—C5—C4 71.8 (4)° and C8—C9—C10—C12 91.9 (4)°], while rings A/B [C6—C5—C11—C1—169.6 (4)°] and E/F [C4—C5—C11—C10—174.4 (4)°] are trans fused. The six-membered rings A, D and E adopt chair, distorted-boat and chair conformations, respectively. The two five-membered rings, C and F, are in envelope conformations, while the seven-membered ring B is found in the twisted-boat form.

Compound (I) contains β -oriented hydroxy and methoxymethyl substituents at C1 and C4, respectively. The tertiary hydroxy group situated at the junction of rings B and D at C8, and the methoxy group at C16 of ring D, are also β -oriented. However, the third hydroxyl group at C14 in ring C is α -oriented with respect to the β -methoxy substituent at C16. The sum of the angles at N1 (337.9 Å) is indicative of sp^3 character.

The packing of (I) is composed of parallel layers stabilized by four $O-H\cdots O$, one $C-H\cdots N$ and three $C-H\cdots O$ interactions (Table 1).

Experimental

Air-dried and powdered aerial parts (2.0 kg) of Delphinium roylei Munz. were extracted exhaustively with *n*-hexane (3×51) followed by ethanol (3×51) over a period of 10 d at room temperature. The ethanolic extract was filtered and concentrated under reduced pressure to afford 166 g of dried mass. The concentrated ethanolic extract was dissolved in distilled water, acidified to pH 2.0 and extracted with dichloromethane to yield a non-alkaloidal mixture (50.0 g). The acidic extract was then basified with 10% KOH (pH 8-10) and extracted with dichloromethane to obtain an alkaloidal fraction (12.0 g). Employment of a repeated column-chromatography procedure using a type 60 (70–230 mesh) silica-gel column (Merck) and a flash (234–300 mesh) silica-gel column (Merck) afforded talatisidine, (I), after elution with petroleum ether-acetone-diethylamine (94.03.1), in 2.31×10^{-3} % yield $(R_f = 0.43, 40\% \text{ acetone-hexane} + \text{a})$ few drops of diethylamine). Compound (I) was recrystallized from petroleum ether-acetone (1:1) as slab-shaped colourless single crys-

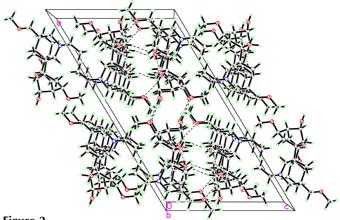


Figure 2

A view, down the b axis, of the molecular packing in (I). Dashed lines indicate hydrogen bonds.

tals. The melting point of (I) is 493 K, similar to the value reported by Pelletier *et al.* (1967).

Crystal data

C23H37NO5	$D_x = 1.265 \text{ Mg m}^{-3}$		
$M_r = 407.54$	Mo $K\alpha$ radiation		
Monoclinic, C2	Cell parameters from 7164		
a = 24.189 (7) Å	reflections		
b = 7.747 (2) Å	$\theta = 1.8 - 25.0^{\circ}$		
c = 13.310 (4) Å	$\mu = 0.09 \text{ mm}^{-1}$		
$\beta = 120.943 (5)^{\circ}$	T = 293 (2) K		
$V = 2139.3 (10) \text{ Å}^3$	Slab, colourless		
Z = 4	$0.44 \times 0.22 \times 0.10 \text{ mm}$		

Data collection

Siemens SMART CCD areadetector diffractometer 2033 independent reflections with $I > 2\sigma(I)$ we scans $R_{\rm int} = 0.025$ Absorption correction: multi-scan (SADABS; Sheldrick, 1996) $h = -27 \rightarrow 28$ $T_{\rm min} = 0.962, T_{\rm max} = 0.991$ $k = -9 \rightarrow 9$ 5739 measured reflections $l = -15 \rightarrow 10$

Refinement

refinement

Refinement on F^2 $w = 1/[\sigma^2(F_o^2) + (0.0384P)^2]$ $R[F^2 > 2\sigma(F^2)] = 0.063$ $wR(F^2) = 0.131$ $where <math>P = (F_o^2 + 2F_c^2)/3$ S = 1.28 $(\Delta/\sigma)_{\rm max} < 0.001$ $\Delta\rho_{\rm max} = 0.22 \ {\rm e}\ {\rm \AA}^{-3}$ $\Delta\rho_{\rm min} = -0.20 \ {\rm e}\ {\rm Å}^{-3}$ H atoms trated by a mixture of

Table 1 Hydrogen-bonding geometry (Å, °).

independent and constrained

D $ H$ $\cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdot \cdot \cdot A$	$D-H\cdots A$
O1-H1O···O2 ⁱ O2-H2O···O3 O2-H2O··O4 ⁱⁱ O3-H3O··O5 ⁱⁱⁱ C2-H2A··N1 C6-H6B··O2 C9-H9···O1 ⁱⁱ	0.82 0.84 (6) 0.84 (6) 0.82 0.97 0.97	2.07 2.37 (6) 2.36 (6) 1.96 2.41 2.57 2.57	2.889 (6) 2.945 (6) 3.130 (5) 2.781 (6) 2.923 (7) 2.911 (7) 3.512 (7)	179 126 (1) 151 (1) 174 113 101 159
C10−H10···O1	0.98	2.40	2.770 (7)	101

Symmetry codes: (i) x, y - 1, z; (ii) $\frac{3}{2} - x, \frac{1}{2} + y, 1 - z$; (iii) 2 - x, y, 1 - z.

organic papers

All H atoms in (I), except for H2O, were positioned geometrically in calculated positions [C—H = 0.96–0.98 Å and O—H = 0.82 Å] and treated as riding atoms, with $U_{\rm iso}({\rm H})$ constrained to be $1.5U_{\rm eq}$ of the carrier atom for methyl and hydroxyl H atoms, and $1.2U_{\rm eq}({\rm parent atom})$ for the remaining H atoms. Both the positional and isotropic displacement parameters of the hydroxyl H atom, H2O, were refined [O—H = 0.82 (4) Å]. The Friedel reflections were merged before the final refinement because of the absence of significant anomalous scattering effects.

Data collection: *SMART* (Siemens, 1996); cell refinement: *SAINT* (Siemens, 1996); data reduction: *SAINT*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 1997); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*, *PARST* (Nardelli, 1995) and *PLATON* (Spek, 2003).

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